Oligonucleotide Analogues with a Nucleobase-Including Backbone

Part 5

2'-Deoxy-8-(hydroxymethyl)adenosine- and 2'-Deoxy-6-(hydroxymethyl)uridine-Derived Phosphoramidites: Synthesis and Incorporation into 14-Mer DNA Strands

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The pairing propensity of new DNA analogues with a phosphinato group between $O-C(3')$ and a newly introduced OCH2 group at C(8) and C(6) of 2'-deoxyadenosine and 2'-deoxyuridine, respectively, was evaluated by force-field calculations and Maruzen model studies. These studies suggest that these analogues may form autonomous pairing systems, and that the incorporation of single modified units into DNA 14mers is compatible with duplex formation. To evaluate the incorporation, we prepared the required phosphoramidites 3 and 4 from 2'-deoxyadenosine and 2'-deoxyuridine, respectively. The phosphoramidite 5 was similarly prepared to estimate the influence of a CH₂OH group at C(8) on the duplex stability. The modified 14-mers $6-9$ were prepared by solid-phase synthesis. Pairing studies show a decrease of the melting temperature by 2.5° for the duplex $13 \cdot 9$, and of $6-8^\circ$ for the duplexes $10 \cdot 6$, $11 \cdot 6$, $13 \cdot 7$, and $14 \cdot 8$, as compared to the unmodified duplexes.

1. Introduction. – We have conceived oligonucleotide analogues with a nucleobaseincluding backbone (Fig. 1, **B**); these compounds should allow us to answer the question of whether the structural differentiation between nucleobase and backbone in $DNA, RNA, and their analogues¹$ is a prerequisite for the formation of stable homoand/or heteroduplexes [6] [7].

We have so far described oligouridine and oligoadenosine analogues that are linked by ethynediyl groups between $C(5')$ and $C(6)$ of uridines, or $C(5')$ and $C(8)$ of adenosines $[6-9]$. In addition to these uncharged analogues that require a stepwise synthesis in solution, we have considered phosphodiester analogues that should be accessible by conventional solid-phase synthesis [10]. Similarly to DNA, they should be water-soluble. These analogues possess phosphinato groups between $O - C(3')$ and a oxymethyl substituent at $C(8)$ or $C(6)$ of neighbouring 2'-deoxyadenosyl or 2'deoxyuridyl units, respectively.

Fig. 2 shows the fully modified analogues $(3' \text{-} 10) dA_n^*$ (1) and $(3' \text{-} 7) dU_n^*$ (2), incorporating the repeating units dA^* and dU^{*2} , and the phosphoramidites 3 and 4

¹⁾ For DNA and RNA analogues that have been prepared in search of antisense-compounds, to investigate autonomous pairing-systems, protein/DNA-interactions, electron-transfer in DNA, and DNA-damage, see $[1 - 5]$.

²) The repeating units of B-DNA are denoted as dA and dT; units that are linked *via* $O(3')$ and $O(10)$, or O(7), respectively, are denoted as dA^* and dU^* , and the dA units CH₂OH-substituted at C(8) as dA^{**} (Fig. 2). The linking mode is indicated by two numerals in parenthesis. A dash, as in $(3'$ -10), indicates a phosphinato group between $O(3')$ and $O(10)$, and indicates the $(3' \rightarrow 10)$ direction of the link. In fully modified analogues, all dA and dT repeating units of DNA are replaced by dA* and dU*, while partially modified analogues correspond to DNA strands in which part of the dA or dT units are replaced by dA^* or dU^* .

Fig. 1. Conventional oligonucleotides with structural separation of the nucleobases and the backbone (A) and oligonucleotides with a nucleobase-including backbone (B)

derived from dA* and dU*, respectively. In this context, we also prepared the phosphoramidite 5 to study the influence of a $C(8)$ -hydroxymethylated dA on the pairing behaviour of DNA oligomers. To avoid unfavourable steric interactions between the Me group at $C(5)$ of thymidine and the backbone connected to $C(6)$, we have replaced thymidine by 2'-deoxyuridine analogues.

We report on force-field calculations of the homoduplex $(3'-10)dA_{14}^* \cdot (7-3')dU_{14}^*$, the heteroduplex $(3' \text{-} 10) dA_{14}^* \cdot (5' \text{-} 3') dT_{14}$, and partially modified duplexes, and describe the synthesis of singly modified tetradecamers containing one dA*, dU*, or dA** unit.

2. Results and Discussion. -1 . *Model Studies and Calculations*. The conformations and the pairing propensity of the above described oligonucleotide analogues ($Fig. 2$) were first evaluated with the help of *Maruzen* models. Models of the fully modified homoduplex $(3'-10)dA_{14}^* \cdot (7-3')dU_{14}^*$, and heteroduplexes $(3'-10)dA_{14}^* \cdot (5'-3')dT_{14}$ and $(3'$ -7)dU^{*}₁₄ $(5'$ -3')dA₁₄, respectively, were constructed on the basis of the results of a conformational analysis of DNA [11], *i.e.*, by assuming a ²E pucker of the carbohydrate moiety, a \pm sc conformation around the P-O bonds, a 'zigzag' conformation of the backbone-chain, and minimal 1,5-repulsions in the single-strands. This led to two plausible helical models for the homoduplex $(3'-10)dA_{14}^* \cdot (7-3')dU_{14}^*$, both characterized by antiparallel strands, but differing in the dihedral angles for the nucleosidic and for the $P-O$ bonds.

Fig. 2. Proposed adenosine and uridine oligomers dA_n^* and dU_n^* , DNA repeating units dA and dT, modified repeating units dA^* , dU^* , and dA^{**} , and phosphoramidites 3, 4, and 5 derived from dA^* , dU^* , and dA^{**} , respectively

In the first model, the nucleobases adopt the anti-conformation with dihedral angles $O(4')-C(1')-N(9)-C(4)$ of ca. -100° in the dA* units and dihedral angles $O(4')-C(1')-N(1)-C(2)$ of ca. -130° in the dU* units, comparable to the corresponding dihedral angles in B-DNA [12]. In the second model, the nucleobases adopt a syn-conformation with dihedral angles $O(4')-C(1')-N(9)-C(4)$ and $O(4') - C(1') - N(1) - C(2)$ of ca. 70 - 80°. Depending on the syn/anti-conformation, the dihedral angles for the P-O bonds are $-sc$ in the first model and $+sc$ in the second one. In both duplexes, the dihedral angles characterizing the torsion about the $C(3')-O(3')$ bond are beween ap and $-ac$. The angles for the $C(8)-C(10)$ and the $C(6)-C(7)$ bonds are ca. 180°, corresponding to an antiperiplanar orientation of N(9) and $O(10)$, and of $N(1)$ and $O(7)$. Both models show a parallel orientation of the

Watson-Crick base pairs at a distance of *ca.* $3.3 - 3.5$ Å, comparable to the distance between base pairs in B-DNA (3.3–3.4 \AA) [13], indicating that these analogues may indeed form stable homoduplexes.

Helical models of the heteroduplexes $(3'-10)dA_{14}^* \cdot (5'-3')dT_{14}$ and $(3'-7)dU_{14}^* \cdot (5'-7)$ $3'$)d A_{14} with antiparallel strands could be constructed only if the nucleobases in the modified strands adopt the *anti*-conformation. The conformations of the dA^* and dU^* units correspond to those of the homoduplexes with an anti-conformation (see above). In the complementary DNA strand, the dA and dTunits adopt conformations similar to those described for dA and dT units in B-DNA [14].

We have also evaluated partially modified analogues. The incorporation of a single dA* or dU* unit into a B-DNA duplex does not considerably change the Watson-Crick base pairing and base stacking, irrespectively of the specific sequence, as long as the nucleobases in the modified units adopt an *anti*-conformation. Again, the conformations of the modified nucleotides are similar to those described for dA^* and dU^* in the models of the homoduplexes characterized by an anti-conformation.

This qualitative analysis formed the basis of force-field calculations (Macromodel V. 4.5, Amber* force-field, gas-phase [15]) to evaluate the potential of the fully and partially modified analogues to form homo- and heteroduplexes similar to DNA. Since the starting geometries were derived from B-DNA, only duplexes with an *anti*conformation of the nucleobases were investigated. The starting geometry of the antiparallel, fully and partially modified duplexes were generated by building the Bhelix of $dA_{14} \cdot dT_{14}$ with the grow-tool of Macromodel, and then replacing some or all nucleotides by the modified nucleotides. The starting conformations of the repeating units dA^* and dU^* were similar to those derived from the *Maruzen* models. After energy minimization (2000 to 3000 iterations), both the homoduplex $(3'-10)dA_{14}^*$ (7-3')dU^{*}₁₄ (*Fig. 3, a*) and the heteroduplex $(3' \text{-} 10) dA^{*}_{14} \cdot (5' \text{-} 3') dT_{14}$ (*Fig. 3, b*) retained a conformation similar to the one initially chosen.

The helical models show an almost parallel arrangement of the nucleobases at a distance of around 3.3 Å, suitable for *Watson-Crick* base pairing and base stacking. Several DNA duplexes partially modified in one strand were similarly investigated. The partially modified B-DNA duplexes containing one, two, or three dA^* or dU^* units with an *anti*-conformation also showed base pairing and base stacking similar to B-DNA. The values of the propeller- and buckle-twists as well as the length of the Hbonds within the Watson-Crick base pairs [16] [17] showed no significant deviation from those in B-DNA.

Neither the position of the modified units in the singly modified strands nor the sequence of these strands appeared (at this level of calculation) to significantly affect the pairing propensity. Among the possible singly modified tetradecamers, we chose to synthesize $6-9$ (Table 1), which possess a modified unit in the center of the chain, flanked by dC and dG units. The choice of 6 was facilitated by an ample supply of the complementary RNA 10^3) that allowed to also study the pairing of 6 with RNA.

2. Synthesis of Oligomers. For the synthesis of the oligomers $6-9$ and $11-14$ (Table 1), we intended to use the standard phosphoramidite chemistry; the (benzoyloxy)methyl group of the required phosphoramidites $3-5$ should not interfere with the

³) We thank *S. Pitsch* for a sample of 5'-r(GAUAGCUCGGAUGC)-3' (10) [18].

(3'-10)dA^{*}₁₄ · (7-3')dU₁₄

 $(3'-10)dA_{14}^* \cdot (5'-3')dT_{14}$

Fig. 3. Amber*-calculated structures of the fully and the semi-modified duplexes $(3'-10)dA_{14}^* \cdot (7-3')dU_{14}^*$ and $(3'-10)dA_{14}^* \cdot (5'-3')dT_{14}$

Sequence		Total yield ^a)	Calculated mass	MALDI-TOF
5'-d(GCATCCGA*GCTATC)-3'	6	87%	4253.8	p.
5'-d(CTATCGA*GCTTACG)-3'	7	87%	4268.8	4268.6
5'-d(CGTAAGCU*CGATAG)-3'	8	80%	4303.8	4308.2
5'-d(CTATCGA**GCTTACG)-3'	9	74%	4268.8	4268.3
5'-d(GATAGCTCGGATGC)-3'	11	91%	4303.8	b١
5'-d(GCATCCGAGCTATC)-3'	12	93%	4223.8	4226.7
5'-d(CGTAAGCTCGATAG)-3'	13	94%	4287.8	4287.4
5'-d(CTATCGAGCTTACG)-3'	14	99%	4238.8	4239.1

Table 1. Oligodeoxynucleotides 6-9 and 11-14, Total Yields, and Molecular Masses

coupling and be readily cleaved under the ammonolytic conditions applied for the deprotection of the oligomers.

To prepare 3, 2'-deoxyadenosine was monobenzoylated in 90% yield by treatment with Me₃SiCl and benzoyl chloride (BzCl) [19] (Scheme 1). Disilylation of the resulting crystalline diol 15 with Et₃SiCl and imidazole in DMF [20] gave 88% of the Et₃Siprotected nucleoside 16. Adenosine derivatives have been $C(8)$ -hydroxymethylated by deprotonation with LDA (= lithium diisopropylamide) or lithium-2,2,6,6-tetramethylpiperidide, addition of ethyl formiate, and reduction with NaBH $_4$ [21]. Subjecting 16 to this protocol yielded only 30% of the desired alcohol 17, while replacing ethyl formiate by DMF increased the yield to ca. 70%. 4,4'-Dimethoxytritylation of 17 gave 18 (89%), which was desilylated to 19 (Bu₄NF in THF; 95%). The regioselective benzoylation of the diol 19 with BzCl and Hünig's base in CH_2Cl_2 , at -10° , as described for 2'deoxyuridine derivatives [22], gave 75% of the monobenzoate 20 that was converted to the phosphoramidite 3 in the usual way [23].

a) Et₃SiCl, imidazole, DMF, 23° ; 88%. b) 1. LDA/THF, -78° ; 2. DMF, -78° ; 3. AcOH, 23° ; 4. NaBH₁/EtOH, 23° ; 70%. c) 4,4'-Dimethoxytrityl chloride (= bis(4-methoxyphenyl)phenylmethyl chloride; DMTrCl), CH₂Cl₂, ${}^{19}P_2$ NEt, 4-(dimethyamino)pyridine (DMAP), 23°; 89% (18), 72% (24). d) Bu₄NF, THF, 23°; 95% (19), 83% (23). e) BzCl, CH₂Cl₂, ⁱPr₂NEt, -10°; 75% (20), 73% (22). f) Ac₂O, Py, 23°; 53%. g) (NCCH₂CH₂O)- $P(^{i}Pr_{2}N)Cl$, $CH_{2}Cl_{2}$, $^{i}Pr_{2}NEt$, 23°; 82% (3), 72% (5).

The phosphoramidite 5, which corresponds to a 3^{\prime} , 5'-linked repeating unit dA**, was prepared in an overall yield of 31% by benzoylation of the alcohol 17, followed by desilylation, 4,4'-dimethoxytritylation at $O(5')$ (72%), and phosphitylation.

The phosphoramidites 3 and 5 are $ca. 3:2$ to 1:1 mixtures of diastereoisomers. At -10° under Ar, they are stable for several months.

The regioselectivity of the benzoylation of 19 was confirmed by N, O -acetylation of the monobenzoate 20. The signals of $H - C(3')$ and both $H - C(5')$ of 20 and of the acetylated product 21 were assigned on the basis of their multiplicity and coupling constants (see Table 5). Compared to H-C(3') of 20 (δ 5.05 ppm), H-C(3') of 21 (Scheme 1) is shifted downfield by 0.57 ppm, while the chemical shift for $H - C(5')$ is hardly affected. The broad d of $HO-C(3')$ at 2.70 ppm (J(H,OH) \approx 1.0 Hz) disappeared upon H/D exchange of 20, while the broad m of $H-C(3')$ collapsed to a dt.

As a rule, the shift from the anti- to the syn-conformation is indicated by a downfield shift of $H_{nro-S}-C(2')$ and $H-C(3')$, and an upfield shift of $C(2')$ [24]. The significant downfield shifts of the signals of $H_{pro-S}-C(2')$ ($\Delta\delta$ = 0.29 – 0.86 ppm) and H-C(3') ($\Delta\delta$ = 0.08 - 0.54 ppm) and the upfield shift of C(2') ($\Delta\delta$ = -0.8 to -4.3 ppm) for the nucleosides $17 - 20$, $22 - 24$, and for the phosphoramidites 3 and 5, as compared to the corresponding signals of 16, evidence that all $C(8)$ -substituted nucleosides adopt predominantly the *syn*-conformation (*Table 2*).

Table 2. ¹H- and ¹³C-NMR Chemical-Shift Differences $\Delta\delta(H-C(2'))$, $\Delta\delta(H-C(3'))$, and $\Delta\delta(C(2'))$ [ppm] for the Nucleosides $3, 5,$ and $17-24$ as Compared to 16 , and Ratio of S/N-Conformers of these Nucleosides, as Deduced from $J(I',2')/J(3',4')$ [Hz]

Nucleoside	$\Delta\delta(H-C(2_{pro-S}^{\prime}))$	$\Delta\delta(H-C(3'))$	$\Delta\delta(C(2))$	J(1',2')	J(3,4')	Ratio S/N
16				6.4	3.8	63/37
17	0.55	0.08	-2.5	6.6	3.7	64/36
18	0.84	0.10	-4.1	6.8	2.8	71/29
19	0.29	0.13	-0.8	9.5	< 1.0	> 90/10
20	0.71	0.42	-3.5	5.6	5.0	53/47
21	0.96	$0.99^{\rm a}$)	-6.6	8.1	3.0	73/27
3	0.86/0.81	0.54/0.45	-2.1	7.4/6.2	5.3/4.5	58/42, 58/42
22	0.84	0.13	-3.0	6.6	3.3	67/33
23	0.39	0.17	-0.8	9.5	< 1.0	> 90/10
24	0.81	0.27	-3.6	6.6	4.1	62/38
5	0.73/0.63	0.35/0.27	-4.3	6.5	$_{p}$	$_{p}$

The ratio $J(1',2'_{pro-5})/J(3',4')$ is considered a measure of the position of the equilibrium between the S- (sugar-pucker ${}_{3}E-{}^{2}T_{1}$) and N-conformers (sugar-pucker ${}^{3}T_{2}$ ${}^{3}T_{4}$) [25]. As expected, the S-conformation is slightly preferred in the O(5')protected nucleosides $16-18$, $20-22$, and 24 , as indicated by the $J(1', 2'_{pro-5})$ (6.2 – 8.1 Hz) and $J(3',4')$ values $(2.8-5.0 \text{ Hz})$ (*Table 2*). In the diols **19** and **23**, $J(1',2'_{pro-5})$ and $J(3',4')$ differ considerably from these values: $J(1',2'_{pro-5}) = 9.5$ and $J(3',4') \le 1.0$ Hz show a clear predominance of the S-conformer of 19 and 23. The strong downfield shift for HO–C(5') of 19 and 23 in CDCl₃ (δ 6.23 and 6.06 ppm, resp.) evidences an intramolecular $O(5')H \cdots N(3)$ H-bond. Such H-bonds are well-documented [7] [26 – 31]. The nine-membered ring formed by the H-bond is rigid $[7] [26] [27] [32]$, as

evidenced by the characteristic $J(5'_{pro-s}, \text{OH}) \leq 1.0$, and $J(5'_{pro-R}, \text{OH}) = 11.0 - 11.2 \text{ Hz}$. Both $J(4',5')$ of 19 and 23 are ≤ 1.5 Hz, indicating a sc arrangement of H–C(4') and both $H-C(5')$, again in agreement with the proposed H-bond. The intramolecular Hbonds in 19 and 23 are confirmed by broad, intensive IR absorptions at 3289 cm⁻¹ and 3286 cm⁻¹ of the diols 19 and 23.

The phosphoramidite 4 was synthesized from 2'-deoxyuridine by silylation to 25 [20], followed by treatment with LDA, DMF, and NaBH₄ (Scheme 2). This yielded only $30 - 35\%$ of the alcohol 26 besides 45% of starting material; 25 and 26 were readily separated by flash chromatography. Increasing or decreasing the amount of LDA and/ or DMF lowered the yield of 26. The unsatisfactory yield is in agreement with results of Miyasaka and co-workers [33], who showed that deprotonation of 2'-deoxyuridine derivatives with LDA occurs in a maximum yield of 52%. BuLi and BuLi/TMEDA $(= N, N, N', N'$ -tetramethylethylenediamine) preferentially deprotonate uridine derivatives at $C(5)$ [33]. Lithium 2,2,6,6-tetramethylpiperidide led to similar results as LDA [21], while the use of LHMDS (= lithium hexamethyldisilazane) did not lead to 26. 4,4'-Dimethoxytritylation of the alcohol 26 to 27, desilylation to the diol 28, and monobenzoylation led to 29, which was converted to the phosphoramidite 4 in the usual way; again, 4 was obtained as a $ca. 1:1$ mixture of diastereoisomers that proved stable at -10° under Ar for several months.

a) 1. LDA/THF, -78° ; 2. DMF, -78° ; 3. AcOH, 23 $^\circ$; 4. NaBH₄/EtOH, 23 $^\circ$; 30% (26), 45% (25). b) DMTrCl, CH_2Cl_2 , ${}^{\rm i}$ Pr₂NEt, DMAP, 23°; 63%. c) Bu₄NF, THF, 23°; 87%. d) BzCl, CH₂Cl₂, ${}^{\rm i}$ Pr₂NEt, -10° ; 65%. e) Ac₂O, Py, 23°; 75%. f) (NCCH₂CH₂O)P(ⁱPr₂N)Cl, CH₂Cl₂, ⁱPr₂NEt, 23°; 72%.

The regioselectivity of the benzoylation was confirmed by H/D exchange and acetylation of 29. The signal of the H $-C(3')$ of 30 (see Table 7) is shifted downfield by 0.69 ppm, while the chemical shift for the two $H - C(5')$ remained practically unaltered. Again, the downfield shifts for $H_{pro-s}-C(2')$ and $H-C(3')$, and the upfield shift for $C(2')$ of both diastereoisomers 4 and of $26 - 30$ (as compared to the corresponding signals of 25) show that the $C(6)$ -substituted nucleosides adopt predominantly the synconformation (Table 3).

Nucleoside	$\Delta\delta(H-C(2_{\text{pro-}S}^{\prime}))$	$\Delta\delta(H-C(3'))$	$\Delta\delta(C(2))$	J(1',2')	J(3,4')	Ratio S/N
25				6.2	3.7	63/37
26	0.50	0.05	-2.8	7.2	4.0	64/36
27	0.82		-3.6	6.2	4.7	57/43
28	0.86	0.27	-3.3	7.5	4.4	63/37
29	0.84	0.33	-3.1	4.1	6.5	39/61
30	1.08	$1.02a$)	-6.3	5.6	5.0	53/47
4	0.84	0.37	-3.9	$b \lambda$	b ₁	b١

Table 3. ¹H- and ¹³C-NMR Chemical-Shift Differences $\Delta\delta(H-C(2'))$, $\Delta\delta(H-C(3'))$, and $\Delta\delta(C(2'))$ [ppm] for the Nucleosides 26-30 and 4 as Compared to 25, and Ratio of S/N-Conformers of these Nucleosides, as Deduced from $J(1',2')/J(3',4')$ [Hz]

The position of the equilibrium between the S- and N-conformers of $25 - 28$ and 30 (*ca.* 60:40) was deduced from the $J(1', 2'_{pro-s})$ and $J(3', 4')$ values (*Table 3*); it is slightly in favour of the S-conformer, while the N-conformer of 29 is slightly preferred. There is no NMR evidence for an intramolecular $O(5')H\cdots O(2)$ H-bond of the diol 28 in CDCl₃. The $J(4'5')$ values (4.1 and 2.8 Hz) of 28 are similar to those of the $O(5')$ protected alcohol 26. These coupling constants evidence that the gauche,gaucheconformation about the $C(4')-C(5')$ bond (essential for an intramolecular H-bond between $HO-C(5')$ and $O(2)$) is less populated in 28 than in the 2'-deoxyadenosinederived diols 19 and 23 ($J(4\prime,5') < 1.5$ Hz) that possess the intramolecular $O(5\prime)H \cdots$ N(3) H-bond.

3. Solid-Phase Synthesis of 14-Mer DNA Strands Containing One dA*, dU*, or dA** Unit, and of the Complementary DNA Strands. Table 1 shows the 14mer DNA strands $11 - 14$ that possess an internal dA or dT unit. Replacement of the internal dA of the unmodified 14mers 12 and 14 by dA^* (derived from 3) led to the modified strands 6 and 7, respectively, while replacement of dA of 14 by dA^{**} (derived from 5) led to 9. The exchange of the internal dT of the 14mer 13 by dU^* (derived from 4) yielded the modified strand 8.

The solid-phase syntheses of the oligomers $6 - 9$ and $11 - 14$ were carried out on a DNA synthesizer and were based on a protocol for the synthesis of pRNA [10]. The coupling time for the attachment of $3-5$ and of the DNA phosphoramidites immediately following the incorporation of $3 - 5$ was arbitrarily set to 30 min; it led to coupling yields between $80 - 92\%$ ⁴) (99% for all other couplings). Longer coupling times did not improve the yields. The 14-mers $6, 7, 9$, and $11 - 14$ were deprotected and cleaved off the solid support by ammonolysis with sat. aqueous $NH₃$ solution/MeOH 1:1 at 50 \degree within 20 h. Prolonged heating of the modified DNA strand 8 (incorporating dU^*) at 50° led to degradation; this strand was deprotected at 35°. The crude products were analyzed by RP-HPLC and showed a composition in agreement with the detritylation assay. The products were purified by RP-HPLC, desalted, and analyzed by RP-HPLC and MALDI-TOF mass spectroscopy. The total yield of the oligomers, as

⁴⁾ Each chain-elongation cycle began with an acid-promoted cleavage of the 4,4'-dimethoxytrityloxy group. The efficiency of each coupling step was calculated by UV determination of the amount of DMTr $(\varepsilon =$ $70000 \cdot \text{mol}^{-1} \cdot \text{cm}^{-1}$).

determined from the detritylation assay, and their molecular masses are shown in Table 1.

4. Pairing Studies. In a first set of measurements, we examined the influence of the incorporation of one dA^* unit into a $DNA \cdot DNA$ duplex or into the DNA strand of a $DNA \cdot RNA$ duplex. The modified 14-mer 6 (Table 1) was hybridized with the complementary 14-mer 11, or its RNA analogue 10, at a concentration of $(2+2)$ µm and a pH of 7 in 0.15m aqueous NaCl (Table 4).

Table 4. Melting Points T_m [$^{\circ}$ C] of 11 12 , 10 12 , 11 \cdot 6, 10 \cdot 6, 13 \cdot 14, 13 \cdot 7, 13 \cdot 9, and 14 \cdot 8 Obtained from Temperature-Dependent UV Spectroscopy

Duplex		$T_{\rm m}$	$\Delta T_{\rm m}$	Strand conc.	Conc. NaCl
5'-d(GATAGCTCGGATGC)-3'	$11 \cdot 12$	57°		$(2+2) \mu M$	0.15M
3'-d(CTATCGAGCCTACG)-5'					
5'-r(GAUAGCUCGGAUGC)-3'	$10 \cdot 12$	53°		$(2+2) \mu M$	0.15M
3'-d(CTATCGAGCCTACG)-5'					
5'-d(GATAGCTCGGATGC)-3'	$11 \cdot 6$	49.5°	-7.5°	$(2+2) \mu M$	0.15M
3'-d(CTATCGA*GCCTACG)-5'					
5'-r(GAUAGCUCGGAUGC)-3'	10.6	45°	-8.0°	$(2+2) \mu M$	0.15M
3'-d(CTATCGA*GCCTACG)-5'					
5'-d(CGTAAGCTCGATAG)-3'	$13 \cdot 14$	50°		$(2+2) \mu M$	0.10 _M
3'-d(GCATTCGAGCTATC)-5'					
5'-d(CGTAAGCTCGATAG)-3'	$13 - 7$	43°	-7.0°	$(2+2) \mu M$	0.10M
3'-d(GCATTCGA*GCTATC)-5'					
5'-d(CGTAAGCTCGATAG)-3'	13.9	47.5°	-2.5°	$(2+2) \mu M$	0.10M
3'-d(GCATTCGA**GCTATC)-5'					
5'-d(CTATCGAGCTTACG)-3'	14.8	44°	-6.0°	$(2+2) \mu M$	0.10M
3'-d(GATAGCU*CGAATGC)-5'					

The modified duplexes show a decrease in melting temperature by 7.5° (11 \cdot 6) and 8.0° (10 \cdot 6) relative to the corresponding non-modified duplexes 11 \cdot 12 and 10 \cdot 12 (see Fig. 4 for the melting curves).

Next, the 14-mer DNA strands $7-9$ incorporating one dA^* , dU^* , or dA^{**} unit were hybridized with the complementary DNA strands 13 or 14 at a concentration of $(2+2)$ μ m and a pH of 7 in 0.1m aqueous NaCl (*Table 4*). The melting curves of the nonmodified duplex 13 \cdot 14, and the modified duplexes 13 \cdot 7 and 14 \cdot 8 are shown in Fig. 5, a.

Again, the duplex 13 \cdot 7 (containing dA*) melted 7° lower than the DNA duplex 13 \cdot **14.** The duplex $14 \cdot 8$ (containing dU^*) showed a depression of the melting temperature by 6° , as compared to the non-modified DNA duplex $13 \cdot 14$.

Pairing of the modified 14mer 9, containing one dA^{**} unit, and the complementary DNA strand 13 yielded a duplex that melted only 2.5° lower than the non-modified DNA duplex $13 \cdot 14$ (Table 4 and Fig. 5, b).

A decrease in melting temperature of $6-8^{\circ}$ for an incorporated dA* or dU* unit indicates a strong destabilization of the DNA duplex. The degree of duplex destabilization caused by dA^* and dU^* is comparable to the degree of duplex destabilization imposed by mismatched base pairs. In the 13mer DNA duplex GCGTACACATGCG CGCATGXGTACGC, for example, the exchange of the matching $X =$ thymidine by adenosine, guanosine, or cytosine leads to a decrease in melting temperature of $5.4 - 13.7^{\circ}$ at a concentration of $(1.5 + 1.5)$ µm and a pH of 7 in

Fig. 4. a) Temperature-dependent UV spectra ('melting curves') of the non-modified and modified DNA \cdot DNA duplexes $11 \cdot 12$ and $11 \cdot 6$ and b) the non-modified and modified DNA \cdot RNA duplexes $10 \cdot 12$ and $10 \cdot 6$

0.1m aqueous NaCl [34]. The decrease by 2.5° of the melting temperature for the duplex 13.9, as compared to $6 - 8^{\circ}$ for the other modified duplexes, evidences the strong influence of the linking mode, rather than just of the introduction of a CH₂OH group, on the duplex destabilisation.

Experimental Part

General. Solvents were distilled before use: THF from K/benzophenone, CH₂Cl₂ from CaH₂. Reactions were run under Ar. Qual. TLC: precoated silica-gel plates (Merck silica gel 60 F_{254}); detection by spraying with 'mostain' (400 ml of 10% aq. H₂SO₄, 20 g of $(NH_4)_{6}M_2O_{24}$ H_2O , 0.4 g of $Ce(SO_4)_{2}$) and heating. Flash chromatography (FC): silica gel Merck 60 (0.04 - 0.063 mm). Optical rotations: 1-dm cell at 25 $^{\circ}$ and 589 nm. FT-IR: $1-2\%$ soln. in the indicated solvent. ¹H₋, ¹³C-, and ³¹P-NMR: at 200, 300, or 400 MHz, 50 or 75 MHz, and

Fig. 5. a) Temperature-dependent UV spectra ('melting curves') of the non-modified and modified DNA \cdot DNA duplexes $13 \cdot 14$, $13 \cdot 7$, and $14 \cdot 8$, and b) $13 \cdot 14$ and $13 \cdot 9$

121 MHz, respectively. MS: fast atom bombardment (FAB), matrix-assisted laser-desorption ionisation (MALDI), high resolution (HR). NOBA: 3-nitrobenzyl alcohol.

 N^6 -Benzoyl-2'-deoxy-3',5'-bis-O-(triethylsilyl)adenosine (16). At 23°, a soln. of 15 [19] (11.6 g, 32.7 mmol) and imidazole (22.3 g, 328 mmol) in DMF (200 ml, distilled from CaH₂) was treated dropwise with Et₃SiCl $(28 \text{ ml}, 165 \text{ mmol})$, stirred for 16 h, treated with MeOH (6 ml) , and evaporated. A soln. of the residue in CH₂Cl₂ was washed with H₂O and brine, dried (Na_2SO_4) , and evaporated. FC (AcOEt/hexane 2:1) gave 16 (16.9 g, 88%). Colourless oil. R_f (AcOEt/hexane 6:1) 0.67. $[\alpha]_D^{25} = -4.8$ ($c = 1.3$, CHCl₃). IR (CHCl₃): 3407w, 2958s, 2912m, 2877m, 1707m, 1612s, 1584m, 1503w, 1455s, 1413m, 1352w, 1329m, 1111m, 1091m, 1071m, 1006m. $1 + \text{NMR}$ (200 MHz, CDCl₃): see *Table 5*; additionally, 9.10 (br. s, NH); 8.38 (s, H – C(8)); 8.10 – 8.00 $(m, 2 \text{ arom. H}); 7.66 - 7.45$ $(m, 3 \text{ arom. H}); 1.10 - 0.85$ $(m, 2 \text{ (MeCH}_2)_3\text{Si}); 0.75 - 0.45$ $(m, 2 \text{ (MeCH}_2)_3\text{Si}).$ ¹³C-NMR (50 MHz, CDCl₃): see *Table 6*; additionally, 165.1 (s, C=O); 134.1 (s); 133.1 (d); 129.1 (d, 2 C); 128.1 $(d, 2 C)$; 7.1, 6.8 (2q, 2 (MeCH₂)₃Si); 5.2, 5.0 (2t, 2 (MeCH₂)₃Si). FAB-MS (NOBA): 584 (64, [M + H]⁺). HR-MALDI-MS: 606.291 ($[M + Na]$ ⁺; calc. 606.291).

	16	17	18	19	20	21	3	22	23	24 5	
$H - C(1')$	6.56	6.47	6.39	6.38	6.29		6.40 6.37, 6.35	6.53	6.56		6.50 6.50, 6.48
$H_{pro-S}-C(2')$	2.69	3.24	3.53	2.98	3.40		3.65 3.60, 3.50	3.53	3.08		3.50 3.42, 3.32
$H_{pro-R}-C(2')$	2.50	2.34	2.20	2.26	2.34		2.38 2.50, 2.48	2.34	2.32		2.42 $2.60 - 2.50$
											$2.50 - 2.42$
$H-C(3')$	4.63	4.71	4.73	4.76	5.05		5.62 5.17, 5.08	4.76	$4.85 - 4.75$		4.90 4.98, 4.90
$H-C(4')$	4.06	4.00	3.92	4.14	4.11		4.30 4.30	4.00	4.22		4.13 4.27
$H - C(5')$	3.84	3.88	3.92	3.97	4.65		4.70 4.76, 4.70	3.89	3.96		$3.40 \text{ } 3.90 - 3.50$
$H-C(5')$	3.72	3.70	3.74	$3.85 - 3.75$	4.60		4.50 4.55, 4.52	3.72	3.77		3.40 $3.90 - 3.50$
$CH-C(8)$	$\overline{}$	5.00	4.55	4.58	4.55		4.55 4.48	5.69	5.65		5.65 5.72
$CH'-C(8)$	\equiv	4.92	4.36	4.41	4.47		4.37 4.43	5.69	5.55		5.55 5.65
$H-C(2)$	8.80	8.75	8.77	8.76	8.60		8.46 8.70	8.80	8.73		8.60 8.63, 8.60
$J(1',2'_{pro-s})$	6.4	6.6	6.8	9.5	5.6	8.1		6.6	9.5	6.6	
$J(1', 2_{pro-R})$	6.4	6.6	6.8	5.4	7.5	6.2		6.6	5.4	6.6	
$J(2_{pro-S}^{'}2_{pro-R}^{'})$	12.9	13.4	13.1	13.3	13.1	14.0		13.3	13.3	13.4	
$J(2_{pro-5}, 3')$	6.4	6.6	6.0	5.4	6.5	6.5		6.5	5.3	6.5	
$J(2_{pro-R}^{\prime}, 3')$	3.7	3.4	3.1	< 1.5	5.6	2.6		3.7	< 1.5	4.1	
J(3,4')	3.8	3.7	2.8	< 1.5	5.0	3.0		3.3	< 1.5	4.1	
J(4,5')	3.8	3.7	5.0	< 1.5	4.4	5.0		4.5	< 1.5	5.0	
J(4',5')	3.8	4.4	6.9	< 1.5	5.0	5.3		5.6	< 1.5	5.0	
J(5',5')	11.2	11.2	13.4	12.4	12.1	11.8		10.8	12.8	< 1.5	
J(CH,CH')	$\overline{}$	14.4	11.8	12.0	12.1	11.8		< 1.5	13.7	13.7	

Table 5. Selected ¹H-NMR Chemical Shifts [ppm] and Coupling Constants [Hz] of 3, 5, and 16–24 in CDCl₃ Solution

Table 6. Selected ¹³C-NMR Chemical Shifts [ppm] of 3, 5, and $16-24$ in CDCl₃ Solution

C(1'), C(4')	C(2')	C(3')	C(5')	$CH2-C(8)$	C(2)	C(4)	C(5)	$C(6)^{a}$	$C(8)^{a}$)
16 85.1, 88.6	41.6	72.1	62.6			153.1 149.6		123.6 151.6	141.6
17 85.1, 88.1	39.1	72.1	62.6	58.1		152.6 149.1		121.6 153.1	155.6
18 85.6, 88.4	37.5	73.0	63.1	59.7		152.6 149.6		122.6 152.9	152.4
19 87.3, 89.6	40.8	73.5	63.5	59.5		151.8 150.0		123.0 152.2	152.7
20 84.3, 84.3	38.1	71.6	64.0	59.6		152.4 149.4		122.3 152.6	152.6
21 82.3, 85.0	35.0	74.7	64.0	59.8		152.2 150.7		128.8 154.6	154.5
3 83.8, 83.7, 84.7, 84.6	37.3	73.3.72.5	63.8	59.7, 59.5		152.6 149.4		122.7 152.6	152.3
22 85.3, 88.1	38.6	72.7	62.6	59.6	153.1 149.1			122.1 152.9	150.2
23 87.0, 90.0	40.8	73.4	63.6	59.2		153.0 149.2		123.3 152.2	150.7
24 85.0, 86.4	38.0	72.9	63.9	59.3		153.1 149.8		122.4 152.6	150.0
5 85.3, 85.1, 85.9, 86.0	37.3	74.5, 73.7	63.8, 63.5	59.6, 59.5		152.7 149.8, 149.6		122.3 152.7 152.4	150.0, 149.9

a) Assignments for $17 - 24$, 3, and 5 may be interchanged.

 N^6 -Benzoyl-2'-deoxy-8-(hydroxymethyl)-3',5'-bis-O-(triethylsilyl)adenosine (17) . At -70° , a soln. of (i-Pr)₂NH (10.3 ml, 72.9 mmol, distilled form CaH₂) in THF (40 ml) was treated dropwise with 1.6m BuLi in hexane (45 ml, 72 mmol), stirred at -70° for 15 min and at 0° for 15 min, and transferred via a syringe to a cooled (-70°) soln. of 16 (8.45 g, 14.4 mmol) in THF (60 ml) within 10 min. The soln. was stirred at -70° for 1 h, treated dropwise with DMF (28 ml, 363 mmol, distilled, from CaH₂) at -70° , stirred for 2.5 h at -70° , treated with AcOH (10 ml), allowed to warm to 23°, diluted with EtOH (100 ml), treated with NaBH₄ (1.76 g, 46 mmol), and stirred for 25 min. After evaporation, the residue was dissolved in CH₂Cl₂, washed with H₂O and brine, dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 1:1) gave 17 (6.2 g, 70%). Colourless foam. R_f $(ACOEt/hexane 2:1) 0.31. [a]_D^{25} = -8.9 (c = 0.65, CHCl_3). IR (CHCl_3): 3405m (br.), 2958s, 2913s, 2878s, 1707s,$ 1655w, 1615s, 1587m, 1526w, 1459s, 1434s, 1345m, 1301w, 1096s. ¹H-NMR (300 MHz, CDCl₃): see *Table 5*;

additionally, 9.23 (br. s, NH); $8.06 - 7.96$ (*m*, 2 arom. H); $7.64 - 7.45$ (*m*, 3 arom. H); 3.80 (br. s, OH); 1.20 - 0.75 $(m, 2(MeCH₂), Si); 0.70 - 0.42 (m, 2(MeCH₂), Si).$ ¹³C-NMR (75 MHz, CDCl₃): see Table 6; additionally, 165.6 $(s, C=O)$; 134.1 (s) ; 133.1 (d) ; 128.9 $(d, 2 C)$; 128.4 $(d, 2 C)$; 6.6 $(q, 2 (MeCH₂)$ ₃Si); 4.9, 4.4 (2t, 2 (MeCH₂)₃Si). FAB-MS (NOBA): 614 (23, $[M + H]$ ⁺). HR-FAB-MS: 614.3187 ($[M + H]$ ⁺; calc. 614.3194).

N⁶-Benzoyl-2'-deoxy-8-[(4,4'-dimethoxytrityloxy)methyl]-3',5'-bis-O-(triethylsilyl)adenosine (18). A soln. of 17 (0.5 g, 0.815 mmol), EtN(i-Pr)₂ (0.55 ml, 3.2 mmol, distilled from CaH₂), and DMAP (50 mg, 0.4 mmol) in CH₂Cl₂ (5 ml) was treated with 4,4'-dimethoxytrityl chloride (1.1 g, 3.25 mmol) at 0° , stirred at 23 $^\circ$ for 16 h, and evaporated. A soln. of the residue in CH₂Cl₂ was washed with H₂O and brine, dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 1:3) gave **18** (670 mg, 89%). Colourless foam. R_f (AcOEt/hexane 1:2) 0.46. $[\alpha]_D^{25} = -6.0$ $(c = 0.8, CHCl₃)$. IR (CHCl₃): 3407w (br.), 2958m, 2912m, 2877m, 2839w, 1706m, 1614s, 1585m, 1509s, 1463m, 1429m, 1414m, 1344m, 1300m, 1116m, 1073m, 1033m. ¹H-NMR (300 MHz, CDCl₃): see *Table 5*; additionally, 9.02 (br. s, NH); 8.06 – 7.96 (m, 2 arom. H); 7.69 – 7.09 (m, 12 arom. H); 6.92 – 6.75 (m, 4 arom. H); 3.77 $(s, 2 \text{ MeO})$; 1.05 – 0.80 $(m, 2 \text{ (MeCH}_2)_3\text{Si})$; 0.70 – 0.42 $(m, 2 \text{ (MeCH}_2)_3\text{Si})$. ¹³C-NMR (75 MHz, CDCl₃): see Table 6; additionally, 162.4 (s, C=O); 159.2 (s, 2 C); 144.5 (s); 135.6 (s); 135.3 (s); 134.4 (s); 132.8 (d); 130.4 – 127.2 (several d); 113.8, 113.7 (2d, 4 C); 87.9 (s, Ph₃C); 55.2 (q, 2 MeO); 6.7, 6.6 (2q, 2 (MeCH₂)₃Si); 4.6, 4.3 (2t, 2) $(MeCH₂)$ ₃Si). FAB-MS (NOBA): 916 (61, $[M+H]^+$), 303 (100, DMTr⁺). HR-MALDI-MS: 938.432 ($[M+H]^+$) Na ⁺; calc. 938.432).

 N^6 -Benzoyl-2'-deoxy-8-[(4,4'-dimethoxytrityloxy)methyl]adenosine (19). A soln. of 18 (660 mg, 0.721 mmol) in THF (6 ml) was treated dropwise at 23° with 1m Bu₄NF \cdot 3H₂O in THF (2.4 ml) for 1.5 h. Evaporation and FC (AcOEt/MeOH 100:1) gave 19 (470 mg, 95%). Colourless foam. R_f (AcOEt/MeOH $100:1)$ 0.39. $[\alpha]_D^{25} = -1.4$ (c = 2.75, CHCl₃). IR (CHCl₃): 3365m (br.), 3286m (br.), 2937m, 2839w, 1709m, 1614s, 1589m, 1509s, 1464m, 1432m, 1347m, 1301m, 1104m, 1062m, 1036m. ¹H-NMR (200 MHz, CDCl₃): see *Table 5*; additionally, 9.20 (br. s, NH); $8.10 - 8.00$ (m, 2 arom. H); $7.68 - 7.16$ (m, 12 arom. H); $6.90 - 6.78$ (m, 4 arom. H); 6.23 (br. d, J = 11.2, HO – C(5')); 3.76 (s, 2 MeO); 2.30 (br. s, HO – C(3')). ¹³C-NMR (50 MHz, CDCl₃): see Table 6; additionally, 165.0 (s, C=O); 159.1 (s, 2 C); 144.5 (s); 135.2 (s); 135.0 (s); 134.0 (s); 133.0 (d); 130.2 -127.3 (several d); 113.7 (d, 4 C); 88.2 (s, Ph₃C); 55.5 (q, 2 MeO). FAB-MS (NOBA): 688 (100, $[M + H]^+$), 303 $(75, DMTr^{+})$. HR-FAB-MS: 688.2769 ($[M + H]^{+}$; calc. 688.2771).

 N^6 ,5'-O-Dibenzoyl-2'-deoxy-8-[(4,4'-dimethoxytrityloxy)methyl]adenosine (20). At -10° (ice/NaCl), a soln. of 19 (454 mg, 0.66 mmol) in CH₂Cl₂ (10 ml) and pyridine (2 ml, 24.9 mmol, distilled from CaH₂) was treated dropwise with a soln. of BzCl (76 μ , 0.66 mmol) in CH₂Cl₂ (2 ml), stirred for 3 h, washed with H₂O (10 ml), dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 4:1) gave 20 (316 mg, 61%). Colourless foam. R_f $(ACOEt/hexane 6:1) 0.55. [a]_D^{25} = -2.2 (c = 0.6, CHCl_3). IR (CHCl_3): 3408m (br.), 2961m, 1711s, 1614s, 1586m,$ 1509s, 1463m, 1430w, 1352w, 1093m, 1070m, 1035m. ¹H-NMR (300 MHz, CDCl₃): see *Table 5*, additionally, 9.00 $(br. s, NH); 8.10-7.90 (m, 4 \text{ arom. H}); 7.71-7.15 (m, 15 \text{ arom. H}); 6.90-6.75 (m, 4 \text{ arom. H}); 3.75 (s, 2 \text{ MeO});$ 2.75 -2.65 (m, HO $-C(3')$). ¹³C-NMR (50 MHz, CDCl₃): see *Table 6*; additionally, 166.8 (s, O $-C=O$); 165.0 $(s, N - C = 0); 159.1 (s, 2 C); 144.3 (s); 135.3 (s); 135.2 (s); 134.2 (s); 133.4 (d); 132.9 (d); 129.9 (s); 130.3 - 127.4$ (several d); 113.8 (d, 4 C); 88.0 (s, Ph₃C); 55.5 (q, 2 MeO). FAB-MS (NOBA): 792 (89, $[M+H]^+$), 303 (100, DMTr⁺). HR-FAB-MS (NOBA): 792.3037 ($[M+H]$ ⁺; calc. 792.3034).

N⁶,3'-O-Diacetyl-N⁶,5'-O-dibenzoyl-2'-deoxy-8-[(4,4'-dimethoxytrityloxy)methyl]adenosine (21). A soln. of 20 (30 mg, 0.044 mmol) in pyridine (1 ml, distilled from CaH₂) was treated with Ac₂O (0.5 ml) and stirred for 16 h at 23°. Evaporation and FC (AcOEt/hexane 1:1) gave 21 (20 mg, 53%). Colourless foam. R_f (AcOEt/ hexane 1:1) 0.33. IR (CHCl₃): 2958w, 1719s, 1607s, 1577m, 1509s, 1449m, 1418w, 1367m, 1347m, 1087m, 1070m, 1035m. ¹H-NMR (300 MHz, CDCl₃): see *Table 5*; additionally, 8.10 – 8.00 (*m*, 2 arom. H); 7.75 – 7.65 $(m, 2 \text{ arom. H})$; 7.60 – 7.20 $(m, 15 \text{ arom. H})$; 6.90 – 6.80 $(m, 4 \text{ arom. H})$; 3.80 $(s, 2 \text{ MeO})$; 2.56 (s, AcN) ; 2.0 (s, AcO). ¹³C-NMR (75 MHz, CDCl₃): see *Table 6*; additionally, 173.1, 172.3 (2s, 2 O–C=O); 170.5, 166.8 (2s, $2 N - C = 0$); 159.3 (s, 2 C); 144.2 (s); 135.2 (s); 135.0 (s); 134.6 (s); 133.4 (d); 133.0 (d); 130.3 (s); 130.0 – 127.5 (several d); 113.8, 113.7 (2d, 4 C); 88.0 (s, Ph₃C); 55.2 (q, 2 MeO); 25.6 (q, Me); 22.2 (q, Me). FAB-MS (NOBA): 876 (14, $[M+H]^+$), 303 (100, DMTr⁺).

N⁶ ,5'-O-Dibenzoyl-2'-deoxy-8-[(4,4'-dimethoxytrityloxy)methyl]adenosine 3'-(2-Cyanoethyl Diisopropylphosphoramidite) (3). A soln. of 20 (80 mg, 0.1 mmol) and $EtN(i-Pr)$, (46 µl, 0.26 mmol, distilled from $CaH₂$) in CH₂Cl₂ (4 ml) was treated with 2-cyanoethyl diisopropylphosphoramidochloridite (31 mg, 0.13 mmol) and stirred for 3 h at 23° . Evaporation and FC (AcOEt/hexane 2:3) gave 3 (81 mg, 82%, 3:2 ratio of 2 diastereoisomers). Colourless foam. R_f (AcOEt/hexane 2:1) 0.25, 0.34 (2 diastereoisomers). $\left[\alpha\right]_D^{25} = -7.1$ (c= 1.63, CHCl3). IR (CHCl3): 3407w (br.), 2970m, 2190w, 1712s, 1614s, 1586m, 1509s, 1463m, 1430w, 1365w, 1352w, 1070m. ¹H-NMR (300 MHz, CDCl₃, 3:2 mixture of 2 diastereoisomers): 8.95 (br. s, NH); 8.70 (s, H–C(2)); 8.05 $-$ 7.90 (m, 4 arom. H); 7.66 $-$ 7.20 (m, 15 arom. H); 6.95 $-$ 6.75 (m, 4 arom. H); 6.37 (dd, J = 7.4, 5.6, 0.4 H),

6.35 (t, $J = 6.0$, 0.6 H) (H-C(1')); 5.17 (ddt, $J = 12.0$, 7.0, 5.3, 0.6 H), 5.08 (ddt, $J = 10.5$, 7.0, 4.5, 0.4 H) $(H-C(3'))$; 4.75 $(dd, J=12.0, 4.5, 0.6 H)$, 4.70 $(dd, J=12.0, 4.5, 0.4 H)$, 4.55 $(dd, J=12.0, 4.5, 0.6 H)$, 4.52 $(dd, J=12.0, 4.5, 0.4$ H) $(2 H-C(5'))$; 4.48, 4.43 $(2d, J=12.1)$ $(CH_2-C(8))$; 4.30 $(dt, J=5.3, 4.5, H-C(4'))$; $3.90 - 3.65$ (m, OCH₂CH₂CN); $3.65 - 3.55$ (m, 0.4 H_{pro-S}-C(2'), (Me₂CH)₂N); 3.76 (s, 2 MeO); 3.50 (ddd, J = 13.0, 7.0, 6.0, 0.6 H_{pro-S} - C(2')); 2.60, 2.53 (2t, J = 6.2, CH₂CN); 2.50 (ddd, J = 13.0, 7.0, 5.6, 0.4 H); 2.48 (ddd, J = 13.0, 7.0, 6.0, 0.6 H) $(H_{m_0,n}-C(2'))$; 1.40 – 1.00 $(m,(Me,CH),N)$. ¹³C-NMR (100 MHz, CDCl₃, 3 : 2 mixture of 2 diastereoisomers): see Table 6; additionally, 166.4, 166.3 (2s, O-C=O); 164.5 (s, N-C=O); 158.9 (s, 2 C); 144.3, 144.2 (2s, 1 C), 135.3, 135.2, 135.1, 134.2, 134.1 (several s, 3 C); 133.2 (d); 132.8 (d); 130.1 (s); 130.1 ± 127.4 (several d); 117.5, 117.4 (2s, CN); 113.6, 113.5, 113.4 (3d, 4 C); 87.9, 87.8 (2s, Ph₃C); 84.7, 84.6 (2dd, ³J(C,P) = 3.7, $C(4')$); 73.3, 72.5 (2dd, ²J(C,P) = 17, C(3')); 58.9, 58.7 (2dt, ²J(C,P) = 20, OCH₂CH₂CN); 55.4 (q, 2 MeO); 43.4, 43.3 (2dd, ²J(C,P) = 7.5, (Me₂CH)₂N); 24.8 – 24.6 (several q, (Me₂CH)₂N); 20.5, 20.4 (2dt, ³J(C,P) = 4.7, OCH₂CH₂CN). ³¹P-NMR (121.5 MHz, CDCl₃): 149.4, 149.3. FAB-MS (NOBA): 992 (29, [M + H]⁺), 303 (100, DMTr+); HR-MALDI-MS: 1014.393 ($[M + Na]$ +; calc. 1014.393).

 N^6 -Benzoyl-8-[(benzoyloxy)methyl]-2'-deoxy-3',5'-bis-O-(triethylsilyl)adenosine (22). At -10° (ice/NaCl), a soln. of 17 (1.68 g, 2.74 mmol) and pyridine (9 ml, 110 mmol, distilled from CaH₂) in CH₂Cl₂ (20 ml) was treated dropwise with BzCl (0.32 ml, 2.74 mmol), stirred for 3 h, washed with H₂O (20 ml), dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 2:3) gave 22 (1.43 g, 73%). Colourless foam. R_f (AcOEt/hexane 1:2) 0.35. $\lbrack a\rbrack_{\rm D}^{25} = 4.3$ (c = 1.3, CHCl₃). IR (CHCl₃): 3325w (br.), 2958s, 2913m, 2877m, 1724s, 1700s, 1615s, 1586m, 1532m, 1461m, 1428m, 1354w, 1109s, 1094s, 1071s. ¹H-NMR (200 MHz, CDCl₃): see *Table 5*; additionally, 9.15 (br. s, NH); 8.14 – 7.95 $(m, 4 \text{ arom. H})$; 7.65 – 7.40 $(m, 6 \text{ arom. H})$; 1.05 – 0.80 $(m, 2 \text{ (MeCH}_2)$ ₃Si); 0.75 – 0.44 $(m, 2 \text{ (MeCH}_2), Si)$. ¹³C-NMR (50 MHz, CDCl₃): see *Table 6*; additionally, 165.9 (s, O–C=O); 164.9 $(s, N-C=O); 134.2 (s); 133.9 (d); 132.9 (d); 130.2 (d, 2 C); 129.4 (s); 129.1 (d, 2 C); 128.8 (d, 2 C); 128.1$ $(d, 2 C)$; 6.8, 6.7 (2q, 2 (MeCH₂)₃Si); 4.8, 4.3 (2t, 2 (MeCH₂)₃Si). FAB-MS (NOBA): 718 (27, [M+H]⁺), 374 (100). HR-MALDI-MS: 740.330 ($[M + Na]$ ⁺; calc. 740.328).

 N^6 -Benzoyl-8-[(benzoyloxy)methyl]-2'-deoxyadenosine (23). At 23°, a soln. of 22 (1.27 g, 1.77 mmol) in THF (40 ml) was treated dropwise with 1m $Bu_4NF \cdot 3 H_2O$ in THF (4.3 ml, 4.3 mmol) and stirred for 1.5 h. Evaporation and FC (AcOEt/MeOH 100:1) gave 23 (720 mg, 83%). Colourless foam. R_f (AcOEt/MeOH $100:1)$ 0.42. $[\alpha]_D^{25} = 27$ (c = 1.13, CHCl₃). IR (CHCl₃): 3402m (br.), 3289 (br.), 2946w, 1726s (br.); 1615s, 1590m, 1532w, 1479s, 1464s, 1452s, 1430s, 1356s, 1106s, 1071s. ¹H-NMR (300 MHz, CDCl₃): see *Table 5*; additionally, 9.26 (br. s, NH); 8.10 – 7.80 (m, 4 arom. H); 7.61 – 7.30 (m, 6 arom. H); 6.06 (dd, J = 11.0, 1.2, HO – C(5')); 3.40 $(d, J = 2.5, HO - C(3'))$. ¹³C-NMR (75 MHz, CDCl₃): see *Table 6*; additionally, 166.2 (s, O-C=O); 165.2 $(s, N-C=O); 134.2$ (d); 134.0 (s); 133.4 (d); 130.3 (d, 2 C); 129.4 (s); 129.4 (d, 2 C); 129.0 (d, 2 C); 128.4 $(d, 2 C)$. FAB-MS (NOBA): 490 (29, $[M + H]^+$), 374 (100).

N⁶-Benzoyl-8-[(benzoyloxy)methyl]-2'-deoxy-5'-O-(4,4'-dimethoxytrityl)adenosine (24). At 0°, a soln. of 23 (678 mg, 1.385 mmol), EtN(i-Pr)₂ (0.55 ml, 3.2 mmol, distilled from CaH₂) and DMAP (50 mg, 0.4 mmol) in pyridine (7 ml, distilled from CaH₂) was treated with $4.4'$ -dimethoxytrityl chloride (0.564 g, 1.67 mmol), stirred for 16 h at 23° , and evaporated. A soln. of the residue in CH₂Cl₂ was washed with H₂O and brine, dried (Na_2SO_4) , and evaporated. FC (AcOEt/hexane 4:1) gave 24 (790 mg, 72%). Colourless foam. R_f (AcOEt/ hexane 4:1) 0.29. $[\alpha]_D^{25} = -2.4$ (c = 1.3, CHCl₃). IR (CHCl₃): 3405w (br.), 3064w, 2959s, 1725s, 1700s, 1614s, 1586s, 1509s, 1464s, 1452s, 1429s, 1353m, 1094s, 1071s, 1036s. ¹H-NMR (300 MHz, CDCl₃): see *Table 5*; additionally, 9.34 (br. s, NH); $8.05 - 7.90$ (m, 4 arom. H); $7.60 - 7.00$ (m, 15 arom. H); 6.80 - 6.65 (m, 4 arom. H); 3.70 (s, 2 MeO). ¹³C-NMR (75 MHz, CDCl₃): see *Table 6*; additionally, 165.9 (s, O–C=O); 164.9 (s, N–C=O); 158.9 (s, 2 C); 144.9 (s); 136.1 (s); 136.0 (s); 133.8 (s); 133.7 (d); 133.1 (d); 129.2 (s); 130.3 - 127.1 (several d); 113.3 (d, 4 C); 86.4 (s, Ph₃C); 55.5 (q, 2 MeO). FAB-MS (NOBA): 792 (4, $[M+H]^+$), 303 (50, DMTr⁺), 374 (100). HR-MALDI-MS: 814.284 ($[M + Na]$ ⁺; calc. 814.285).

N⁶ -Benzoyl-8-[(benzoyloxy)methyl]-2'-deoxy-5'-O-(4,4'-dimethoxytrityl)adenosine 3'-(2-Cyanoethyl Diisopropylphosphoramidite) (5). At 23°, a soln. of 24 (188 mg, 0.237 mmol) and EtN(i-Pr)₂ (0.11 ml, 0.62 mmol, distilled from CaH₂) in CH₂Cl₂ (12 ml) was treated with 2-cyanoethyl diisopropylphosphoramidochloridite $(53 \mu, 0.238 \text{ mmol})$ and stirred for 3 h. Evaporation and FC (AcOEt/hexane 2:3) gave 5 (169 mg, 72%, 1:1) mixture of 2 diastereoisomers). Colourless foam. R_f (AcOEt/hexane 1:2) 0.26, 0.33 (2 diastereoisomers). $\lbrack a\rbrack_{\rm D}^{25} = -10.3$ (c = 1.15, CHCl₃). IR (CHCl₃): 3405w (br.), 2969m, 2935m, 2200w, 1725s, 1700s, 1614s, 1586m, 1509s, 1463s, 1428m, 1365m, 1353m, 1094s, 1071s, 1036s. ¹H-NMR (300 MHz, CDCl₃, 1:1 mixture of 2 diastereoisomers): 9.00 (br. s, NH); 8.63, 8.60 (2s, H $-C(2)$); 8.08 (d, 2 arom. H); 8.00 (d, 2 arom. H); 7.60 – 7.10 $(m, 15 \text{ arom. H})$; 6.80 – 6.65 $(m, 4 \text{ arom. H})$; 6.50, 6.48 (2t, J = 6.5, H – C(1')); 5.72, 5.65 (2d, J = 13.6), $CH_2-C(8)$; 4.98 (ddt, J = 14.7, 5.6, 4.5, 0.5 H), 4.90 (ddt, J = 13.2, 5.6, 2.8, 0.5 H) (H-C(3')); 4.27 (dt, J = 5.6, 4.9, H $-C(4')$); 3.90 - 3.50 (m, 2H $-C(5')$, (Me₂CH)₂N, OCH₂CH₂CN); 3.68 (s, 2 MeO); 3.42 (ddd, J

12.0, 6.5, 4.5, 0.5 H), 3.32 (ddd, J = 10.5, 6.5, 2.8, 0.5 H) (H_{pro-S}-C(2')); 2.60 - 2.40 (m, H_{pro-R}-C(2')); 2.57 (td, J = 6.3, 2.1); 2.45 (t, J = 6.4, CH₂CN); 1.25 – 1.00 (m, (Me₂CH)₂N). ¹³C-NMR (100 MHz, CDCl₃, 1:1 mixture of 2 diastereoisomers): see Table 6; additionally, 165.7 (s, O $-C=O$); 164.5 (s, N $-C=O$); 158.55, 158.5, 158.45 (3s, 2 C); 144.8, 144.7 (2s, 1 C); 136.2, 136.1, 135.9, 134.0 (several s, 3 C); 133.7 (d); 132.8 (d); 130.2 – 126.8 (s, several d); 117.6, 117.56 (2s, CN); 113.1 (d, 4 C); 86.35, 86.3 (2s, Ph₃C); 74.3 (dd, ²J(C,P) = 16.8), 73.5 $(dd, {}^2J(C,P) = 17.9$ $(C(3'))$; 58.6, 58.5 $(2dt, {}^2J(C,P) = 16.2$, OCH₂CH₂CN); 55.3, 55.2 $(2q, 2 \text{ MeO})$; 43.35 $(2dd, {}^{2}J(C, P) = 12.1, (Me_{2}CH)_{2}N); 24.75 - 24.55$ (several q, $(Me_{2}CH)_{2}N); 20.5, 20.3$ $(2dt, {}^{3}J(C, P) = 7.1,$ OCH₂CH₂CN). ³¹P-NMR (121.5 MHz, CDCl₃): 149.5, 149.2. FAB-MS (NOBA): 992 (23, [M+H]⁺), 303 (100, DMTr⁺). HR-MALDI-MS: 1014.393 ($[M + Na]$ ⁺; calc. 1014.393).

 $2'-Deoxy-3', 5'-bis-O-(triethylsilyl)uridine (25)$. A soln. of 2'-deoxyuridine (10 g, 43.9 mmol) and imidazole $(29.7 g, 436 mmol)$ in DMF $(150 ml)$ was treated dropwise with Et₃SiCl $(37 ml, 218 mmol)$, stirred for 16 h at 23° , treated with MeOH (8 ml), and evaporated. A soln. of the residue in CH₂Cl₂ was washed with H₂O and brine, dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 1:1) gave 25 (18.9 g, 94%). Colourless oil. R_1 $(ACOEt/hexane 1:1) 0.67. [a]_D^{25} = 31.8 (c = 0.4, CHCl₃). IR (CHCl₃): 3392m (br.), 2958s, 2913m, 2878s, 1690s,$ 1634w, 1460s, 1415m, 1388m, 1116s, 1084m, 1067m, 1006m. ¹H-NMR (200 MHz, CDCl₃): see *Table 7*, additionally, 8.67 (br. s, NH); 8.00 $(d, J = 8.3, H - C(6))$; 0.98 - 0.84 $(m, 2(MeCH₂)$ ₃Si); 0.70 - 0.42 $(m, 2)$ $(MeCH₂)$ ₃Si). ¹³C-NMR (75 MHz, CDCl₃): see *Table 8*; additionally, 6.6, 6.4 (2q, 2 (MeCH₂)₃Si); 4.6, 4.2 (2t, 2) $(MeCH₂)₃Si)$. FAB-MS (NOBA): 457 (88, $[M+H]$ ⁺). HR-MALDI-MS: 479.236 ($[M+Na]$ ⁺; calc. 479.237).

2'-Deoxy-6-(hydroxymethyl)-3',5'-bis-O-(triethylsilyl)uridine (26). At -70° , a soln. of (i-Pr)₂NH (10.3 ml, 72.9 mmol, distilled from CaH2) in THF (40 ml) was treated dropwise with 1.6m BuLi in hexane (45 ml, 72 mmol), stirred at -70° for 15 min and at 0° for 15 min, and transferred via syringe to a cooled (-70°) soln. of 25 (6.6 g, 14.4 mmol) in THF (60 ml) within 10 min. The soln. was stirred at -70° for 1 h, treated with DMF (28 ml, 363 mmol, distilled from CaH₂), stirred for 2.5 h, treated with AcOH (10 ml), allowed to warm to 23^o, diluted with EtOH (100 ml), treated with NaBH₄ (1.76 g, 46 mmol), and stirred for 25 min. After evaporation, the residue was dissolved in CH₂Cl₂, washed with H₂O and brine, and dried (Na₂SO₄). Evaporation and FC (AcOEt/hexane 1:1) gave 26 (2.1 g, 30%) and 25 (3 g, 46%). Colourless foam. R_f (AcOEt/hexane 1:1) 0.27. $\lbrack a\rbrack_5^2 = 22.5$ (c = 0.4, CHCl₃). IR (CHCl₃): 3392m (br.), 2958s, 2913m, 2878m, 1697s, 1639s, 1458m, 1412m,

	25	26	27	28	29	30	4
$H - C(1')$	6.33	6.20	5.75	5.75	5.74	5.80	$5.78 - 5.65$
$H_{pro-S}-C(2')$	2.10	2.60	2.92	2.96	2.94	3.18	2.94
$H_{pro-R}-C(2')$	2.33	2.20	2.02	2.07	2.18	2.13	$2.30 - 2.15$
$H - C(3')$	4.45	4.50	4.38	4.72	4.78	5.47	$4.90 - 4.72$
$H - C(4')$	3.95	3.83	$3.8 - 3.6$	3.80	3.92	4.15	$4.12 - 4.00$
$H-C(5')$	3.90	3.90	$3.8 - 3.6$	3.82	4.61	4.58	4.70, 4.60
$H - C(5')$	3.75	3.80	$3.8 - 3.6$	3.76	4.50	4.50	4.42, 4.38
$CH-C(6)$		4.55	3.97	4.00	4.00	3.95	4.00
$CH'-C(6)$		4.45	3.91	3.93	3.95	3.95	4.00
$H-C(5)$	5.70	5.78	5.73	5.66	5.70	5.68	$5.78 - 5.65$
$J(1',2'_{pro-S})$	6.2	7.2	6.2	7.5	4.1	5.6	
$J(1',2'_{pro-R})$	6.2	7.2	8.1	7.8	8.7	8.1	
$J(2_{pro-S}^{\prime}, 2_{pro-R}^{\prime})$	13.3	13.2	13.1	13.7	13.7	13.8	
$J(2_{pro-5}^{\prime}, 3')$	6.2	7.5	7.5	7.4	8.1	8.0	
$J(2_{pro-R}^{\prime}, 3')$	4.3	3.7	4.7	4.4	6.5	4.7	
J(3,4')	3.7	4.0	4.7	4.4	6.5	5.0	
J(4,5')	2.2	2.8	$^{a})$	2.8	4.1	4.8	
J(4,5')	2.5	3.8	a)	4.1	6.2	6.8	
J(5',5')	11.2	10.6	a)	12.8	11.8	11.8	
J(CH,CH')		14.0	12.9	14.3	13.4	< 1.5	

Table 7. Selected ¹H-NMR Chemical Shifts [ppm] and Coupling Constants [Hz] of 4, and $25-30$ in CDCl₃ Solution

	C(1'), C(4')	C(2')	C(3')	$C(5')$, $CH_2-C(6)$	C(2)	C(4)	C(5)	C(6)
25	85.3, 88.0	42.0	71.2	$62.0 -$	150.6	163.9	102.0	140.5
26	85.8, 87.9	39.2	71.8	63.0, 61.0	150.8	164.4	101.9	156.6
27	86.6, 88.3	38.4	72.6	63.8, 62.7	150.3	163.6	102.6	153.6
28	87.6, 87.2	38.7	71.2	62.8, 62.3	151.6	163.3	103.7	153.1
29	84.9, 86.4	38.9	72.1	65.3, 62.6	150.9	163.4	103.3	153.2
30	82.4, 86.8	35.7	74.8	64.9, 62.9	150.3	163.0	103.5	152.9
4	84.2, 83.7	38.1	73.8	64.7, 62.6	150.3	163.6	103.0	153.1
	86.7, 86.3		73.0					153.0

Table 8. Selected ¹³C-NMR Chemical Shifts [ppm] of 4, and $25-30$ in CDCl₃ Solution

1373m, 1338m, 1084s, 1042m, 1006 m. ¹H-NMR (300 MHz, CDCl₃): see *Table 7*; additionally, 8.20 (br. s, NH); 3.40 (br. s, OH); 1.05 – 0.84 (m, 2 (MeCH₂) \cdot Si); 0.70 – 0.50 (m, 2 (MeCH₂) \cdot Si), ¹³C-NMR (50 MHz, CDCl₃); see Table 8; additionally, 6.7 (q, 2 (MeCH₂)₃Si); 4.7, 4.3 (2t, 2 (MeCH₂)₃Si). FAB-MS (NOBA): 487 (48, [M + H]⁺).

 $2'-Deoxy-6-[(4,4'-dimethoxytritylow)methyl-3',5'-bis-O-(triethylsily)uridine (27). A soln. of 26 (3.7 g,$ 7.6 mmol), EtN(i-Pr)₂ (5.2 ml, 30.2 mmol, distilled from CaH₂), and DMAP (50 mg, 0.4 mmol) in CH₂Cl₂ (60 ml) was treated with 4,4'-dimethoxytrityl chloride (10.3 g, 30.4 mmol) at 0° , stirred at 23° for 16 h, and evaporated. A soln. of the residue in CH₂Cl₂ was washed with H₂O and brine, dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 1:3) gave 27 (3.8 g, 63%). Colourless foam. R_f (AcOEt/hexane 1:3) 0.28. $[\alpha]_D^{25} = 9.9$ (c= 1.45, CHCl3). IR (CHCl3): 3390w (br.), 2958s, 2938m, 2912m, 2877s, 1691s, 1608m, 1584w, 1510s, 1464s, 1445m, $1414m$, $1379m$, $1342w$, $1071s$, $1035s$, $1006s$. $H\text{-NMR}$ (200 MHz, CDCl₃): see *Table 7*; additionally, 8.30 (br. s, NH); 7.50 – 7.20 $(m, 9 \text{ arom. H})$; 6.95 – 6.80 $(m, 4 \text{ arom. H})$; 3.76 $(s, 2 \text{ MeO})$); 1.05 – 0.84 $(m, 2 \text{ (MeCH}_2)$ ₃Si); 0.75 $-$ 0.50 (m, 2 (MeCH₂)₃Si). ¹³C-NMR (50 MHz, CDCl₃): see Table 8; additionally, 159.2 (s, 2 C); 144.2 (s); 135.3 (s); 135.0 (s); 130.1 – 127.5 (several d); 113.8, 113.7 (2d, 4 C); 88.0 (s, Ph₃C); 55.3 (q, 2 MeO); 6.7 (q, 2 $(MeCH₂)$ ₃Si); 4.6, 4.2 (2t, 2 (MeCH₂)₃Si). FAB-MS (NOBA): 789 (2, $[M+H]$ ⁺), 303 (100, DMTr⁺); HR-FAB-MS: 788.3883 (M⁺; calc. 788.3888).

 $2'-Deoxy-6-(4,4'-dimethoxytrityloxy)methyl/uridine (28)$. A soln. of 27 (3.55 g, 4.5 mmol) in THF (60 ml) was treated dropwise at 23° with 1m Bu₄NF \cdot 3 H₂O in THF (10.9 ml, 10.9 mmol) and stirred for 1.5 h. Evaporation and FC (AcOEt/MeOH 100 : 1) gave 28 (2.2 g, 87%). Colourless foam. R_f (AcOEt/MeOH 100 : 1) 0.56. $\left[\alpha\right]_D^{25} = 3.8 \left(c = 0.42, \text{CHCl}_3\right)$. IR (CHCl₃): 3390m (br.), 2959m, 2840m, 1694s, 1608m, 1584m, 1509s, 1464m, $1445m$, $1411m$, $1382m$, $1303m$, $1064m$, $1035s$. 1 H-NMR (300 MHz, CDCl₃): see *Table 7*; additionally, 7.50 – 7.20 $(m, 9 \text{ arom. H})$; 6.86 (d, 4 arom. H); 3.80 (s, 2 MeO). ¹³C-NMR (50 MHz, CDCl₃): see *Table 8*; additionally, 159.4 (s, 2 C); 144.1 (s); 135.2 (s); 135.0 (s); 130.2 – 127.7 (several d); 113.9, 113.8 (2d, 4 C); 88.3 (s, Ph₃C); 55.5 $(q, 2 \text{ MeO})$. FAB-MS (NOBA): 560 (13, M^+), 303 (100, DMTr⁺).

5'-O-Benzoyl-2'-deoxy-6-[(4,4'-dimethoxytrityloxy)methyl]uridine (29). At -10° (ice/NaCl), a soln. of 28 (2.2 g, 2.92 mmol) in CH₂Cl₂ (100 ml) and pyridine (23 ml, 284 mmol, distilled from CaH₂) was treated dropwise with a soln. of BzCl (0.56 ml, 4.86 mmol) in CH₂Cl₂ (10 ml), stirred at -10° for 3 h, washed with H₂O (60 ml), dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 4:1) gave 29 (1.7 g, 65%) and 28 (0.5 g, 23%). Colourless foam. R_f (AcOEt/hexane 6:1) 0.44. [a] $_0^{25}$ = 8.9 (c = 1.32, CHCl₃). IR (CHCl₃): 3391*m* (br.), 3186w (br.), 2959*m*, 2936m, 2840w, 1694s, 1608m, 1584w, 1509s, 1464m, 1452m, 1409m, 1381m, 1316m, 1095s, 1070s, 1036s. ¹ H-NMR (300 MHz, CDCl₃): see *Table 7*; additionally, 8.80 (br. s, NH); 8.12 – 8.00 (*m*, 2 arom. H); 7.64 – 7.20 (*m*, 12 arom. H); 6.94 – 6.83 (m, 4 arom. H); 3.80 (s, 2 MeO)); 2.30 (br. d, $J = 4.0$, OH). ¹³C-NMR (50 MHz, CDCl₃): see Table 8; additionally, 167.2 $(s, O - C = 0)$; 159.4 $(s, 2 C)$; 144.1 (s) ; 135.2 (s) ; 135.0 (s) ; 133.4 (d) ; 130.3 – 127.7 $(s, several d)$; 113.9, 113.8 (2d, 4 C); 88.1 (s, Ph_1C) ; 55.5 $(q, 2 \text{ MeO})$. FAB-MS (NOBA): 664 (4, M^+), 303 (100, DMTr⁺). HR-FAB-MS (NOBA): 664.2417 (M^+ ; calc. 664.2421).

3'-O-Acetyl-5'-O-benzoyl-2'-deoxy-6-[(4,4'-dimethoxytrityloxy)methyl]uridine (30). At 23°, a soln. of 29 (20 mg, 0.03 mmol) in pyridine (1 ml, distilled from CaH₂) was treated with Ac₂O (0.5 ml) and stirred for 16 h. Evaporation and FC (AcOEt/hexane 1:1) gave 30 (16 mg, 75%). Colourless foam. $\left[a\right]_D^{25} = 2.3$ ($c = 0.92$, CHCl₃). IR (CHCl3): 3390w (br.), 2936w, 1722s, 1692s, 1608m, 1510m, 1462m, 1452m, 1378m, 1095m, 1070m, 1036m. $1H\text{-NMR}$ (300 MHz, CDCl₃): see *Table 7*; additionally, 9.00 (br. s, NH); 8.10 – 8.00 (*m*, 2 arom. H); 7.60 – 7.20 $(m, 12 \text{ arom. H})$; 6.90 – 6.80 $(m, 4 \text{ arom. H})$; 3.80 $(s, 2 \text{ MeO})$; 1.95 (s, Me) . ¹³C-NMR (75 MHz, CDCl₃): see Table 8; additionally, 170.5, 166.7 (2s, 2 C=O); 159.4 (s, 2 C); 144.2 (s); 135.2 (s); 135.0 (s); 133.4 (d); 130.3 (s); 130.2 – 127.7 (several d); 113.8, 113.7 (2d, 4 C); 88.2 (s, Ph₃C); 55.5 (q, 2 MeO); 20.9 (q, Me).

5'-O-Benzoyl-2'-deoxy-6-[(4,4'-dimethoxytrityloxy)methyl]uridine 3'-(2-Cyanoethyl) Diisopropylphosphoramidite) (4). A soln. of 29 (100 mg, 0.15 mmol) and $EtN(i-Pr)_2$ (90 µl, 0.52 mmol, distilled from CaH₂) in CH₂Cl₂ (4 ml) was treated with 2-cyanoethyl diisopropylphosphoramidochloridite (34 μ l, 0.152 mmol) and stirred at 23 $^{\circ}$ for 3 h. Evaporation and FC (AcOEt/hexane 2:3) gave 4 (93 mg, 72%, 2:1 ratio of 2 diastereoisomers). Colourless foam. R_f (AcOEt/hexane 1:2) 0.24, 0.33 (2 diastereoisomers). IR (CHCl₃): 3390w (br.), 2969w, 2935w, 1718s, 1694s, 1608w, 1509m, 1463m, 1414w, 1379m, 1366w, 1094m, 1070m, 1035m. ¹ H-NMR (300 MHz, CDCl₃, 2:1 ratio of 2 diastereoisomers): $8.10 - 8.00$ (m, 2 arom. H); 7.60 – 7.20 (m, 12 arom. H); 6.90 -6.80 (m, 4 arom. H); 5.78 -5.65 (m, H $-C(5)$, H $-C(1')$); 4.90 -4.72 (m, H $-C(3')$); 4.70 (dd, J = 11.8, 3.1, 0.67 H), 4.60 (dd, J = 12.0, 3.7, 0.33 H), 4.42 (dd, J = 11.8, 3.4, 0.67 H), 4.38 (dd, J = 12.0, 3.4, 0.33 H) $(2 H-C(5'))$; 4.12 - 4.00 (m, H – C(4')); 4.00 (br. s, CH₂ – C(6)); 3.90 – 3.40 (m, (Me₂CH)₂N, OCH₂CH₂CN); 3.76 (s, 2 MeO); 2.94 (ddd, J = 13.0, 8.4, 4.7, $H_{pro-5} - C(2')$); 2.58, 2.57 (2t, J = 6.3, CH₂CN); 2.30 - 2.15 (m, $H_{pro}R-C(2')$); 1.10 – 1.00 (m, $(Me_2CH)_2N$). ¹³C-NMR (75 MHz, CDCl₃, 2 : 1 ratio of 2 diastereoisomers): see Table 8; additionally, 166.7 (s, O – C=O); 159.3 (s, 2 C), 144.1 (s); 135.25, 135.2, 135.0 (3s, 2 C); 133.3 (d); 130.3 (s); 130.2 – 127.6 (several d); 117.8 (s, CN); 113.9 (d, 4 C); 88.1 (s, Ph₃C); 84.2, 83.7 (2dd, ³J(C,P) = 4.5, C(4')); 73.8 (dd, ²J(C,P) = 19), 73.0 (dd, ²J(C,P) = 17) (C(3')); 59.0, 58.6 (2dt, ²J(C,P) = 20, OCH₂CH₂CN); 55.4 $(q, 2 \text{ MeO})$; 43.5, 43.2 (2dd, ²J(C,P) = 12, (Me₂CH)₂N); 24.9 – 24.4 (several q, (Me₂CH)₂N); 20.6 – 20.1 (2dt, 3*I*(C P) – 4.6. OCH.CH.CN)³ 3P NMR (121.5 MHz, CDCL); 149.9, 149.7. FARMS (NORA); 865.(1.5 M + ${}^{3}J(C,P) = 4.6$, OCH₂CH₂CN). ${}^{3}I$ P-NMR (121.5 MHz, CDCl₃): 149.9, 149.7. FAB-MS (NOBA): 865 (1, [*M* + H]⁺), 303 (100, DMTr⁺). HR-MALDI-MS: 887.340 ($[M + Na]$ ⁺; calc. 887.340).

Oligonucleotide Synthesis. Oligonucleotide syntheses were performed on a Pharmacia Gene Assembler on a 1.3-umol scale. The commercial phosphoramidites and the CPG solid supports were obtained from Glen Research. Solvents and reagents were prepared essentially according to the protocol for the synthesis of pRNA [10]. Detritylation was accomplished within 2 min by 3% Cl₂CHCOOH in (CH₂Cl)₂. Couplings (0.16 ml of 0.1m phosphoramidite soln. $+0.36$ ml of 0.35 M 1H-tetrazole/0.15M 5-(4-nitrophenyl)-1H-tetrazole soln. in MeCN, or 0.36 ml 0.25m 1-(benzylthio)-1H-tetrazole soln. in MeCN, resp.) were performed within $6-10$ min (DNAphosphoramidites coupled on unmodified nucleotides), or 30 min (modified phosphoramidites, and DNAphosphoramidites coupled on modified nucleotides, resp.). Capping and oxidation was accomplished under standard conditions [35].

Deprotection and Purification of Oligonucleotides. Removal of the protecting groups and detachment from the solid support was effected in conc. aq. NH₃ soln./MeOH 1 : 1 (2.5 ml) at 50 $^{\circ}$ in 20 h (oligomers 6, 7, 9, and 11 – 14) or at 35 \degree in 5 h (oligomer 8 containing dU*; see Results and Discussion). After filtration and evaporation, the crude oligomers were dissolved in aq. buffer soln. $(0.1M HOAc, 0.1M Et₃N)$, purified by RP-HPLC, and desalted by RP-HPLC. Their composition was confirmed by MALDI-TOF-MS.

REFERENCES

- [1] E. Uhlmann, A. Peyman, Chem. Rev. 1990, 90, 543.
- [2] Y. S. Sanghvi, P. D. Cook, Carbohydrate Modifications in Antisense Research, Washington, DC, 1994, p. 232.
- [3] G. L. Verdine, H. Huang, R. Chopra, S. C. Harrison, Science 1998, 282, 1669.
- [4] D. Ly, L. Sanii, G. B. Schuster, J. Am. Chem. Soc. 1999, 121, 9400.
- [5] R. Epple, E. U. Wallenborn, T. Carell, J. Am. Chem. Soc. 1997, 119, 7440.
- [6] S. Eppacher, N. Solladié, B. Bernet, A. Vasella, *Helv. Chim. Acta* 2000, 83, 1311.
- [7] H. Gunji, A. Vasella, Helv. Chim. Acta 2000, 83, 1331.
- [8] H. Gunji, A. Vasella, Helv. Chim. Acta 2000, 83, 2975.
- [9] H. Gunji, A. Vasella, Helv. Chim. Acta 2000, 83, 3229.
- [10] S. Pitsch, S. Wendeborn, B. Jaun, A. Eschenmoser, Helv. Chim. Acta 1993, 76, 2161.
- [11] A. Eschenmoser, M. Dobler, *Helv. Chim. Acta* 1992, 75, 218.
- [12] W. Saenger, 'Principles of Nucleic Acid Structure', Springer Verlag, 1984, p. 268.
- [13] W. Saenger, 'Principles of Nucleic Acid Structure', Springer Verlag, 1984, p. 226.
- [14] W. Saenger, 'Principles of Nucleic Acid Structure', Springer Verlag, 1984, p. 232.
- [15] F. Mohamadi, N. G. J. Richards, W. C. Guida, R. Liskamp, M. Lipton, C. Caufield, G. Chang, T. Hendrickson, W. C. Still, J. Comput. Chem. 1990, 11, 440.
- [16] W. Saenger, 'Principles of Nucleic Acid Structure', Springer Verlag, 1984, p. 26.
- [17] W. Saenger, 'Principles of Nucleic Acid Structure', Springer Verlag, 1984, pp. 116–126.
- [18] X. Wu, Swiss Federal Institute of Technology (ETH), Diss. ETH No. 13567, 2000.

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- [19] L. Gaffney, R. A. Jones, G. S. Ti, J. Am. Chem. Soc. 1982, 104, 1316.
- [20] E. J. Corey, A. Venkateswarlu, J. Am. Chem. Soc. 1972, 94, 6190.
- [21] H. Hayakawa, K. Haraguchi, H. Tanaka, T. Miyasaka, Chem. Pharm. Bull. 1987, 35, 72.
- [22] J. Yamashita, H. Matsumoto, K. Kobayashi, K. Noguchi, M. Yasumoto, T. Ueda, Chem. Pharm. Bull. 1989, 37, 2287.
- [23] S. Pitsch, *Helv. Chim. Acta* 1997, 80, 2286.
- [24] L. Dudycz, R. Stolarski, R. Pless, D. Shugar, Z. Naturforsch. C: Biosci. 1978, 34, 359.
- [25] D. B. Davies, 'Conformations of Nucleosides and Nucleotides', in 'Progress in NMR Spectroscopy', 1978, Vol. 12, p. 135.
- [26] D. Plochocka, A. Rabczenko, D. B. Davies, Biochim. Biophys. Acta 1977, 476, 1.
- [27] L. H. Koole, H. Deboer, J. W. Dehaan, A. G. Haasnoot, P. Vandael, H. M. Buck, J. Chem. Soc., Chem. Commun. 1986, 362.
- [28] T. Ueda, Y. Nomoto, A. Matsuda, Chem. Pharm. Bull. 1985, 33, 3263.
- [29] S. T. Rao, M. Sundaralingam, J. Am. Chem. Soc. 1970, 92, 4963.
- [30] V. Swaminathan, M. Sundaralingam, J. B. Chattopadhyaya, C. B. Reese, Acta Crystallogr., Sect. B 1980, 36, 828.
- [31] F. Seela, H. Debelak, H. Reuter, G. Kastner, I. A. Mikhailopulo, Nucleosides Nucleotides 1998, 17, 729.
- [32] D. Gani, A. W. Johnson, J. Chem. Soc., Perkin Trans. 1 1982, 1197.
- [33] H. Tanaka, H. Hayakawa, S. Iijima, K. Haraguchi, T. Miyasaka, Tetrahedron 1985, 41, 861.
- [34] M. Berger, A. Ogawa, D. McMinn, Y. Wu, P. Schultz, F. Romesberg, Angew. Chem., Int. Ed., 2000, 39, 2940.
- [35] Pharmacia, 'User Manual for Gene Assembler Plus'.

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